Annals of Occupational Hygiene

- Oxford Journals
- Medicine
- The Annals of Occupational Hygiene
- Volume 53, Number 3
- Pp. 289-296
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The Annals of Occupational Hygiene...

Annals of Occupational Hygiene Advance Access originally published online on March 11, 2009

Annals of Occupational Hygiene 2009 53(3):289-296; doi:10.1093/annhyg/mep009

This Article

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Permeation of Hair Dye Ingredients, p-Phenylenediamine and Aminophenol Isomers, through Protective Gloves

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Skin irritation and contact allergies are skin disorders common to hairdressers. The predominant oxidative hair dye components, such as *p*-phenylenediamine (PPD) and aminophenol isomers, can cause contact dermatitis. Use of protective gloves can prevent dermal contact with skin irritants. This study investigates the permeation behaviors of *p*-aminophenol (PAP), *m*-aminophenol (MAP), *o*-aminophenol (OAP) and PPD in single and mixed challenge solutions with disposable natural rubber latex (NRL) gloves, disposable polyvinylchloride (PVC) gloves

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and neoprene (NP) gloves. The challenge solutions were 4% PPD (w/v), 3% OAP (w/v), 2% PAP (w/v) and 2% MAP (w/v) in ethanol or 12% hydrogen peroxide



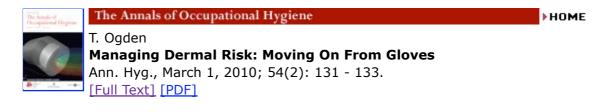
solutions. The cocktail solutions of the four chemicals were also tested. An American Society for Testing and Materials type permeation cell, ethanol liquid collection and gas chromatography-flame ionization detection of samples taken from the collection medium every 10 min facilitated determination of breakthrough times (BTs), cumulative permeated masses and steady-state permeation rates (SSPRs). Experiments were 4 h long for the NRL and PVC gloves and 8 h for NP gloves. No chemicals tested broke through the NP gloves when exposed for 8 h. In the ethanol solution, PPD and OAP started breaking through the PVC gloves at 40 min. The SSPRs of PVC gloves were higher than those for NRL gloves in all challenge conditions for both single chemicals and mixtures. No tested chemicals in hydrogen peroxide solutions permeated the gloves during the 4-h tests. The chemical composition of the challenge solution was a main effecter of BTs and SSPRs for the NRL glove. For disposable PVC gloves, the main factors of BTs were molecular size [molar volume (MV)] and polarity ($log K_{ow}$), and the primary factors of SSPRs were concentration, MV and $\log K_{ow}$. In conclusion, disposable NRL gloves and disposable PVC gloves should not be used repeatedly for handling the hair dye products. Hydrogen peroxide did not accelerate chemical breakthrough. The compositions of the challenge solutions and physical and chemical properties (MV and $\log K_{ow}$) affected permeation behaviors for different gloves.

Keywords: aminophenols • chemical-protective gloves • hair dyes • permeation • p-phenylenediamine

Received July 30, 2008; in final form February 10, 2009



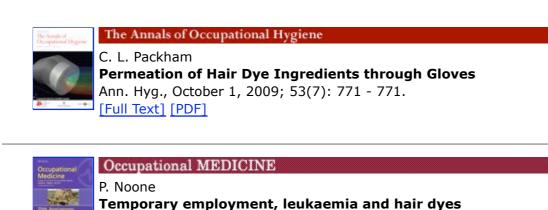
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Online ISSN 1475-3162 - Print ISSN 0003-4878

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